CALS-based Systemic Metrological Studies

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Abstract—The paper describes CALS-based systemic metrological studies related to the data reduction process using the example of photometric analysis for the amines content in an aqueous mixture of dimethylethanolamine, cyclohexylamine and morpholine. The results obtained in the course of the comparison show that the divergence between the results of different laboratories fall inside the $CD_{0.95}$ norm. In other words, results of all laboratories are acceptable. This conclusion together with intermediate data was entered in the Data Reduction Process subcategory of the CALS project.

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Computer-aided development of specifications, one of the key parts of a product documentation package, requires the use of the state-of-the-art information-management systems. The most advanced computer-support system for this is the Continuous Acquisition and Life Cycle Support (CALS) technology providing continuous information-guided support for the product lifecycle [1].

Experience gained in the process of implementing various self-contained information systems showed that to support all the stages of the product lifecycle [2], an enterprise or a group of enterprises (a virtual enterprise) must have an integrated information environment (IIE) involving different information technologies. Lifecycle (LC) begins with the inception of the idea of a new product and ends with the disposal of the manufactured product at the end of its useful life. In between, the lifecycle stages include marketing engineering, study. process design (PD), manufacturing itself, aftersales service, and use of the product.

Every stage of the lifecycle is associated with its specific objectives, and those in charge strive to reach the stated aims with a maximum efficiency. At the engineering and PD stages the aim is to meet the quality requirements to the product with a preset degree of its reliability and minimization of material and time expenditures, which is requisite for success in

the competitive struggle in a market-driven economy. The notion of effectiveness involves not only cost reduction and saving of engineering and production time, but also the convenience of introducing new products and the reduction of the cost of the future use of the product. The cost of the future use is particularly important for such sophisticated items as products of organic and inorganic chemistry.

The IIE approach and the integration of information technology at the LC stages underlie the initiative known in the USA under the CALS acronym [3]. The initiative was first introduced by the United States Department of Defense as a measure to reduce the cost of information interaction between public agencies and commercial entities involved in the delivery of arms and military technologies.

The gist of the CALS initiative consists in using information-support principles and methods at all the stages of the product LC. The result of it is uniform methods of managing processes and interactions of all the parties involved in the cycle: product customers (including governmental agencies and services), product suppliers (manufacturers), and maintenance crews. These principles and methods are implemented according to international management and cooperation standards primarily by means of electronic data interchange.

The IIE, which is the basis and the core of the CALS initiative, is actually a distributed data storage facility residing in the network computer system which encompasses, ideally, all the services and subdivisions of the enterprise involved in the product LC processes. There is a unified system of rules of presentation, storage and sharing of information in the IIE. According to these rules, information processes supporting the product LC at all its stages take place within the IIE. Implemented in this case is the core principle of the CALS initiative: once captured at any stage of the LC, all information becomes available to every person involved in that stage, as well as in all other stages (provided those persons are entitled to use such information). With such an arrangement there will be no unnecessary duplication, conversion, or unauthorized alteration of data as well as no mistakes caused by these procedures; and it also reduces the cost of labor, time expenditures, and offers financial saving [4].

Within the IIE, using application software such as CAE/CAD/CAM, PDM, MRP/ERP, SCM, etc., the information is created, stored and shared among the persons involved in the LC [3].

The main feature of the CALS initiative distinguishing it from any other is that its basic principles and technologies are implemented (wholly or partially) over the LC of any product irrespective of its designation or physical characteristics [5]. Among the technologies that may be considered basic is managing of projects, product configuration, integrated information environment, quality, workflow, and of production and organizational set-ups [2].

In the broad sense of the term, CALS is a methodology creating a common information space, which provides, for industrial products, interaction of all industrial automation systems (AS) and integration of all persons involved in the life cycle of the product in question. In this sense, the subject of CALS is methods and tools for interaction of both various automation systems and of their subsystems, including the provision of all types of their support. Actually, alongside CALS, there is another term, Product Lifecycle Management (PLM), which is increasingly used today in the same sense by leading manufacturers of AS.

In the narrow sense of the term, CALS is a technology for merging various ASs with their specific linguistic, informational, programmed, mathematical,

methodological, equipment-support and organizational types of support. Linguistic support includes data languages and formats for industrial goods and processes used for representing and sharing information between ASs and their subsystems at various stages of the products' LC. Information support consists of industrial goods information databases used by various systems in the process of manufacturing, maintenance and disposal of products. Information support also includes series of international and national CALS-related standards and specifications.

CALS software consists of program complexes designed to support common information space of the products' LC stages. These are, first of all, document-management and document-flow systems, PDM systems, design tool for interactive electronic technical manuals and some other instruments.

Mathematical support of CALS includes methods and algorithms for creating and using models of interaction between various systems within the CALS technologies. The first to be mentioned among these methods are simulation modeling of complex systems, processes planning methods and resource assignment. Methodological support of CALS includes techniques of execution of such processes as concurrent design and production, structuring of complex objects, their functional and information modeling, object-oriented design, and creation of software ontologies.

Equipment support of CALS includes hardware facilities for capturing, storing, processing and visualisation of data in the course of information support of products. Interaction of subdivisions of virtual enterprises and systems supporting various stages of the LC is effected via data transmission lines and switching devices. Widely used are opportunities offered by the Internet and Web technologies. However, the hardware used is not CALS-specific.

Organizational support of CALS includes a range of documents, agreements and manuals specifying roles and responsibilities of parties involved in the life cycle of industrial goods.

Following are some quantitative estimates of effectiveness of CALS implementation in the USA production sector [6]: saving of direct design expenses 10–30%, saving of product development time 40–60%, time saving for introduction of products to the market 25–75%, reduction in defectives and in number of engineering changes 23–73%, saving of expenses

for preparation of engineering documentation up to 40%, and saving of expenses for preparation of maintenance documentation up to 30%.

Currently, most Russian enterprises and agencies do not realize the importance of implementing, as soon as possible, CALS technologies in industrial production. Information technologies (ITs) are used by Russian industry, primarily, for addressing isolated problems of designing, process engineering, preproduction, operations management, etc. By contrast, at some of the leading research and development establishments and production facilities CALS technologies were successfully used for a long time. However, the ITs used were, as a rule, inconsistent with international CALS standards. The main reason for it is that the complexity of transition from the use of ITs at isolated stages of products' LC to working in an IIE encompassing all LC stages is underestimated [7].

International practice shows that it takes five to seven years, from the time when the necessity of introducing CALS technologies in the manufacturing industry has been realized, to obtain tangible results of it. Therefore, it is fair to assume that in Russia, it will also take a lot of time to reach effective use of the CALS initiative. Should the introduction of CALS technologies in Russian industry be unreasonably delayed, the country may lose external market of knowledge-intensive products and have difficulties in partaking in industrial cooperation.

The major problem concerning ITs that has to be addressed as soon as possible is working out electronic description of products in the course of their engineering, process design and manufacturing. This problem is of prime importance for enterprises selling licenses for industrial technology of knowledge-intensive products, first of all, military hardware.

Within the CALS initiative, it is very important for the enterprises to actually embrace paperless technologies in engineering, manufacturing and servicing the products. To do it, an appropriate normative framework regulating the use of e-documents and digital signature has to be in place. An important activity area is developing guidance manuals and software solutions on integrated logistics support of knowledge-intensive products.

Today, CALS technologies are being implemented actively and successfully mostly at enterprises of the military-industrial complex. In chemicals sector, the use of the CALS initiative is described primarily in publications by researchers from the Federal State Unitary Enterprise (FSUE) State Scientific-Research Institute of Chemical Reagents and High Purity Chemical Substances (known under the IREA abbreviation) and R&D Centre "Fine Chemicals" (RDC FC). Over the last 15 years, the two establishments solved a number of most topical problems of the chemicals sector using CALS-specific program complex.

In the nanotechnology realm [8, 9], the problem of recovery of ultra pure silicon in a plasma-chemical reactor under cryogenic quenching was considered. Within the framework of the information-driven CALS project, a master scheme (application protocol), Reference Data for Design (RDFD), for the engineering of an apparatus for the production of nanomaterials was created. According to chemical sector standards, RDFD includes 17 obligatory sections. All the sections, together with their subsection, are entered in the CALS project.

The production of highly purified material is one of the priority research areas at IREA and RDC FC [10, 11]. The development of information-driven CALS projects was undertaken along several lines. Within the framework of the concept under discussion, a set of studies was conducted concerning the engineering of industrial production of highly purified mineral acids. To solve the problem, we devised a CALS project for the production of highly purified mineral acids using *Reference Data for Design* application protocol together with the computer-aided quality management CALS system to analytically monitor the abovementioned range of acids [12, 13].

To computerize, following the ISO 10303 STEP information standard, the development of specifications for highly purified acids, we created a pertinent CALS project [14] with sections of specifications prescribe under interstate standard GOST 2.114-95. In the CALS system, these sections are represented in the form of 9 information categories and subcategories of various levels featuring distinct hierarchic structure. The first stage of the CALS system included two most important, for the analytical chemists, information categories: Technical Specifications (#1) and Testing Methods (#5).

Another line of our research in the realm of preparation of highly purified chemical substances is flexible technologies for the preparation of aliphatic hydrocarbons and petroleum spirits of reactive qualification [15]. To address the problem, we also created database patterns for issuing reference data for design and used the CALS project with all the 17 sections of RDFD and the relevant documentation entered in it.

Within the framework of creating new light-weight heat-protection materials (HPMs), we developed a research-and-information complex consisting of CALS projects targeted at analyzing raw material base, information and patent sources [16]. We have comprehensively studied and structured available current information on the aforesaid materials. For instance, in the CALS project analyzing patent sources, patents we examined were arranged into countries-of-patentee groups and according to duration of patents. The CALS project for analyzing raw-material base, where the data on the currently used for the production of HPMs raw materials was structured, had been used in the developing of new compositions of HPM.

Within the framework of organic synthesis, we carried out two work packages. The first one was related to the development of CALS projects in module technology for preparation of a range of ionic liquids [17]. To study the mechanism of synthesis of ionic liquids (ILs) containing 1,3-dialkylimidazolium cation, which we use for the synthesis of a range of a combined materials, promising three-stage laboratory-scale plant was devised. The plan includes 6 main process stages for the production of all three liquids. Basing on experimental research, a pilot plant for the synthesis of ILs had been engineered. A CALS project for a temporary process guide for the preparation of a range of ILs with all 14 main sections included in it and with all the data on the manufacturing process (detail description of all main devices and processing lines, and with a drawing of the flow chart) was also created. High emphasis was placed on the instrumentation of the rectification and chemical purification stages.

Scientific studies under consideration at the second work package were related to the preparation of phosphonoacetic acid (PAA) derivatives [18]. The method for the preparation of 3-methyl PAA ester is based on the reaction of phase-transfer PH alkylation in an organic phase/solid alkali metal carbonate system. Theoretical and experimental research allowed us to develop an information CALS system structured

along three lines: Catalytic Synthesis, Separation of the Target Product, and Product Quality Assessment. Data entered in the subcategories of the system reflect the most important properties of the chemicals and processes under consideration (effect of various parameters on the yield of the target product, method of processing, and quality level of the resulting substance).

In the Ecology focus area, a range of works on waste management in the phosphorus industry had been completed [19]. We undertook a study on industrial phosphorus-containing waste treatment in Russia and Kazakhstan. An effective method of industrial phosphorus-containing waste management is to use the waste for the production of a road-building material (asphalt concrete). However, because phosphorous slag and phosphogypsum pose health and environmental risks, their compositions and radioactivity levels have to be studied. With this end in view, we entered in the CALS system analytical research data for rare and heavy metals, radioactive elements, etc. An informational subsystem involving findings on the concentrations of radionuclides and chemical pollutants in phosphorous slag and phosphogypsum, along with the environmental pollution indices, was developed.

Alongside the issue of industrial phosphorus-containing waste management, we dealt with CALS projects involving the environmental impact of deicing agents (DAs) [20, 21]. To handle that problem, we developed a project with all the necessary information entered in it: list of environmental objects subject to DA impact (topsoil, water bodies, vegetation, and the atmosphere), classes and names of substances occurring is DAs, and quality parameter that were used in the environmental impact assessment. The developed CALS system contains all necessary data on the assessment method, instrument used, and format of the output documentation for every quality parameter.

Aside from the information system for monitoring the environmental impact of DAs, CALS projects involving information on the DAs and asphalt—concrete binders used in Moscow, as well as on the new DA with a higher concentration of sodium formate and the new type of binder, that are being created at the IREA, were developed [20]. The information entered in the projects includes the quality parameters of the materials under investigation, quality assessment methods, and instruments used.

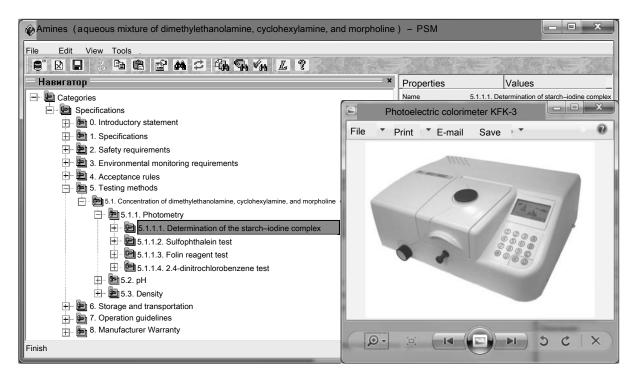


Fig. 1. CALS project specifications: testing methods–main substance–photometry–determination of the starch–iodine complex (photoelectric colorimeter KFK-3).

Basing on the ISO 10303 STEP information standard, a master scheme (application protocol) for specifications on commercial chemicals was previously developed [14]. In the development of the master scheme, the information structure was based on the interstate standard GOST 2.114-95 [22] prescribing the general rules for the design, description, formatting, coordination, and approval of product specifications. The standard database structure (Fig. 1) consists of an introductory statement and 8 main upper-level information categories: (1) specifications; (2) safety requirements; (3) environmental monitoring requirements; (4) acceptance rules; (5) testing methods; (6) storage and transportation; (7) operation guidelines; and (8) manufacturer warranty.

Category 5: Testing Methods lays down the techniques, methods and modes of control (tests, measurements, and analyses) of parameters, norms, and characteristic of and requirements to products that should be monitored under the provisions of Category 4: Acceptance Rules. For every control method (test, measurement, and analysis) the following should be established, depending on the particular character of the method: sampling techniques, equipment, materials and chemical reagents, etc.; mode of preparation for

the control (test, measurement, or analysis), method of control (test, measurement, or analysis), and data processing. All these blocks are subcategories of the CALS system.

In the present paper, we put emphasis upon the Data Processing subcategory. In this respect, we will consider the CALS project under Category 5: *Testing Methods*, using the example of photometric analysis of amines in an aqueous mixture of dimethylethanolamine, cyclohexylamine, and morpholine (subcategory 5.1). The samples are also tested for two quality parameters: pH (subcategory 5.2) and density (subcategory 5.3).

The main photometric methods of analysis (subcategory 5.1.1) for the main substance in the aboventention of a starch-iodine complex formed by the oxidation of potassium iodide with chlorine-containing compounds resulting from the reaction between amines and hypochlorite (subcategory 5.1.1.1); sulfophthalein (bromocresol purple) test, as a result of which yellow chloroform-extractable products are formed (subcategory 5.1.1.2); Folin reagent (sodium 1,2-naphthoquinone-4-sulfonate) test with the subsequent extraction of colored compounds with chloroform and

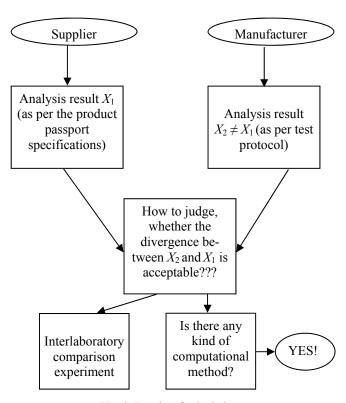


Fig. 2. Results of calculations.

photometric measurements (subcategory 5.1.1.3); and 2.4-dinitrochlorobenzene test with the subsequent extraction of colored products and photometric measurements (subcategory 5.1.1.4).

The photometric analysis for the starch–iodine complex by means of a KFK-3 photoelectric colorimeter proved to be the most appropriate method in terms of its sensitivity (the detection limit declared in the procedure is 1 µg per sample).

The CALS project of the *Testing Methods* category contains in the first-level subcategory *Determination of Dimethylethanolamine, Cyclohexylamine, and Morpholine* the subcategory *Data Processing*, where, within the description of the requirements to data processing, computing formulas, calculation accuracy, and data rounding up, as well as the permissible divergences of the results of parallel calculations are given (Fig. 2).

The quality of the output product is one of the main performance criteria of any enterprise, and the properties of raw and other materials play a significant role in it. Today, the manufacturer assesses the quality of raw materials by comparing the results of their incoming analysis with the characteristics specified by the supplier in the accompanying documents (passport specifications), and very often a large divergence between the supplier's and consumer's data is observed.

The divergences between the results of different laboratories should be assessed at the stage of the development of the test procedures (for instance, in the technical specifications for the product) after an interlaboratory comparison experiment. But for economic and management reasons such experiment is quite difficult to perform (no less than 20 or 30 laboratories should be involved). Therefore, it is preferable to asses such divergences using the characteristics of the test methods, derived computationally—experimentally in the course of certification of the methods, rather than through an interlaboratory comparison experiment.

The national guidelines [23] give formalized methods for the computational assessment of repeatability, accuracy, and precision (expanded uncertainty) of the test methods which cannot be certified through an interlaboratory comparison experiment. However, the document does not give any recommendations as to repeatability assessment in this case. The method of assessment of the critical difference $CD_{0.95}$ [24], exemplified in the present paper by the method of the determination of amines in an aqueous mixture of dimethylethanolamine, cyclohexylamine and morpholine may fill in the gap.

To prepare the method for certification, the results of the photometric determination on a KFK-3 instrument of the concentration of amines in the condensate within the range 0.5–20 mg/dm³ were used. Basing on the results of the analysis, the integral error Δ of the measurement procedure and the critical difference $CD_{0.95}$ at every calibration point were determined.

The experimental data were obtained in a series of parallel determinations for two concentration ranges: 0.5–2.0 mg/dm³ and 2.0–20.0 mg/dm³. The results for the first concentration range are presented in Table 1 and for the second, in Table 2.

Based on the results of parallel determinations, the following was calculated: the arithmetic mean $Y_{i, \text{mean}}$ for each concentration within the range, using the formula

$$Y_{i, \text{ mean}} = \frac{\sum Y_{ij}}{n}$$

where n is the number of parallel determinations of every concentration within the range (in our case, n = 3); and the absolute value of the sampling variance $S_{i,abs}^2$, using the formula

Table 1. Analysis results for the concentration range 0.5–2.0 mg/dm³

| Amine concentration, mg/dm³ | Optical density (parallel determinations) | | |
|-----------------------------|---|----------|----------|
| X_i | Y_{i1} | Y_{i2} | Y_{i3} |
| 0.50 | 0.141 | 0.131 | 0.122 |
| 1.00 | 0.363 | 0.375 | 0.352 |
| 1.50 | 0.557 | 0.519 | 0.534 |
| 2.00 | 0.712 | 0.660 | 0.708 |

$$S_{i, \text{ abs}}^2 = \frac{\sum (Y_{ij} - Y_{i, \text{ mean}})^2}{n-1}.$$

The experimental data were tested for outliers. To this end, the purpose ratio $|Y_{ij \max(\min)} - Y_{i, \max}|/S_{i, abs}$ was calculated, where $|Y_{ij \max(\min)} - Y_{i, \max}|$ is the maximum divergence from the mean values in parallel determination (for each concentration).

The results of the tests for outliers, entered in the CLAS project, are presented in Tables 3 and 4.

The resulting data proved not to exceed the tabulated value of 1.15 (at n = 3 and P = 0.95), indicating a lack of outliers.

We also corroborated the hypothesis of homogeneity of the absolute sampling variance according to the Cochran criterion and calculated the estimated mean square weighted deviation.

Basing on the experimental data, the linear approximation accuracy and Fisher variance ratio (*F*) were calculated. The result of this was that the hypothesis of linearity in the case of amines concentrations in the

Table 3. Results of tests for outliers: amine concentrations in the condensate within the range 0.5–2.0 mg/dm³

| Run no. | X ₁ , mg/dm³ | $Y_{i,\text{mean}}$ | $S_{ m abs}$ | $\frac{ Y_{ij \max(\min)} - Y_{i, \text{mean}} }{S_{i, \text{abs}}}$ |
|------------|-------------------------|---------------------|--------------|--|
| 1 | 0.5 | 0.141 | 0.010 | 1.02 |
| 2 | 1.0 | 0.363 | 0.012 | 1.01 |
| 3 | 1.5 | 0.557 | 0.019 | 1.06 |
| 4 | 2.0 | 0.712 | 0.029 | 1.15 |

Table 2. Analysis results for the concentration range 2.0–20.0 mg/dm³

| Amine concentration, mg/dm ³ | Optical density (parallel determinations) | | |
|---|---|-------|-------|
| X_i | Y_{i1} Y_{i2} Y_{i3} | | |
| 2.0 | 0.066 | 0.048 | 0.060 |
| 5.0 | 0.129 | 0.112 | 0.102 |
| 10.0 | 0.230 | 0.215 | 0.210 |
| 15.0 | 0.308 | 0.318 | 0.315 |
| 20.0 | 0.410 | 0.425 | 0.420 |

condensate within the range of 0.5 to 2.0 mg/dm³ was ruled out ($F_{\rm exp} > F_{\rm tab}$); by contrast, in the case of amine concentrations in the condensate within the range of 2.0 to 20.0 mg/dm³, the linearity hypothesis was accepted ($F_{\rm exp} < F_{\rm tab}$).

The residual systematic component of the error of the analysis was calculated by the formula:

$$\theta = 1.1\sqrt{\theta_{\mu}^2 + \theta_{\text{calib}}^2},$$

where $\Sigma \theta_{\mu}^2$ is the sum of the residual errors of the preparation of solutions for calibration and analysis and $\theta_{\text{calib}}^2 = 3S_{YI}^2$ is the calibration (or approximation) error calculated statistically by the least-squares method [25].

To calculate the resulting systematic error, the errors for every point within the range of amine concentrations in the condensate were determined.

Whereas the inequality $8 < \theta/(SX_{mean}\epsilon) < 0.8$ is true for either range of amine concentrations in the condensate under consideration, the range of the resulting error of analysis was determined by constructing

Table 4. Results of tests for outliers: amine concentrations in the condensate within the range 2.0–20.0 mg/dm³

| Run no. | X_1 , mg/dm ³ | $Y_{i, \text{mean}}$ | $S_{ m abs}$ | $\frac{ Y_{ij \max(\min)} - Y_{i, \text{mean}} }{S_{i, \text{abs}}}$ |
|------------|----------------------------|----------------------|--------------|--|
| 1 | 2.0 | 0.058 | 0.009 | 1.09 |
| 2 | 5.0 | 0.114 | 0.014 | 1.07 |
| 3 | 10.0 | 0.218 | 0.010 | 1.12 |
| 4 | 15.0 | 0.314 | 0.005 | 1.10 |
| 5 | 20.0 | 0.418 | 0.008 | 1.09 |

Table 5. Integral systematic errors of the determination of amine concentrations in the condensate within the range 0.5–2.0 mg/dm³

| Concentration, mg/dm ³ | $\frac{\theta}{SX_{\text{mean}}\epsilon}$ | K | S_{Σ} | Δ |
|--------------------------------------|---|------|--------------|----|
| 0.5 | 3.30 | 1.92 | 8.240314 | 16 |
| 1.0 | 1.15 | 1.94 | 2.867430 | 6 |
| 1.5 | 0.81 | 1.96 | 2.019092 | 4 |
| 2.0 | 0.89 | 1.95 | 2.220604 | 4 |

composed distribution of random and residual systematic errors, considered to be accidental variables, by the formula $\Delta = KS_{\Sigma}$, where K is the coefficient reflecting the random-to-residual systematic error ratio, and S_{Σ} , estimated integral rms deviation of the analysis result.

Table 6. Integral systematic errors of the determination of amine concentrations in the condensate within the range 2.0–20.0 mg/dm³

| Concentration, mg/dm³ | $\frac{\theta}{SX_{\text{mean}}\epsilon}$ | K | S_{Σ} | Δ |
|-----------------------|---|------|--------------|---|
| 2.0 | 8.88 | _ | _ | 9 |
| 5.0 | 4.94 | 1.91 | 2.498344 | 5 |
| 10.0 | 3.28 | 1.92 | 1.657823 | 3 |
| 15.0 | 2.76 | 1.92 | 1.395820 | 3 |
| 20.0 | 2.56 | 1.92 | 1.291739 | 3 |

The calculated integral errors of analysis are presented in Tables 5 and 6.

The estimated acceptability of the calculated interlaboratory divergence between the results of analysis can be illustrated by a comparison of the

Table 7. Calculated mean values, sampling variances, and Cochran criteria

| Name of laboratory | Amine concentration in the condensate, mg/dm ³ | Analysis results | | $X_{i, \mathrm{mean}}$ | $S_{i,\mathrm{abs}}^2$ | σ |
|--------------------|---|------------------|----------|-----------------------------|--------------------------------|-----------|
| | | X_{i1} | X_{i2} | $\Lambda_{i, \text{ mean}}$ | $\mathcal{O}_{i,\mathrm{abs}}$ | g_{max} |
| IREA | 0.5 | 0.45 | 0.50 | 0.48 | 0.001250 | 0.8571 |
| | 1.0 | 1.00 | 1.10 | 1.05 | 0.005000 | |
| | 1.5 | 1.45 | 1.50 | 1.48 | 0.001250 | |
| | 2.0 | 2.00 | 2.30 | 2.15 | 0.045000 | |
| MPEI | 0.5 | 0.50 | 0.48 | 0.49 | 0.000200 | 0.7576 |
| | 1.0 | 1.10 | 1.10 | 1.10 | 0.000000 | |
| | 1.5 | 1.45 | 1.43 | 1.44 | 0.000200 | |
| | 2.0 | 2.00 | 1.95 | 1.98 | 0.001250 | |
| Travers | 0.5 | 0.48 | 0.45 | 0.47 | 0.000450 | 0.6289 |
| | 1.0 | 1.05 | 1.10 | 1.08 | 0.001250 | |
| | 1.5 | 1.45 | 1.55 | 1.50 | 0.005000 | |
| | 2.0 | 1.90 | 1.95 | 1.93 | 0.001250 | |
| IREA | 5.0 | 4.50 | 4.60 | 4.60 | 0.005000 | 0.5000 |
| | 10.0 | 9.70 | 9.60 | 9.70 | 0.005000 | |
| MPEI | 5.0 | 4.80 | 4.60 | 4.70 | 0.020000 | 0.8000 |
| | 10.0 | 9.50 | 9.60 | 9.60 | 0.005000 | |
| Travers | 5.0 | 4.40 | 4.50 | 4.50 | 0.005000 | 0.5000 |
| | 10.0 | 9.60 | 9.70 | 9.70 | 0.005000 | |

Table 8. Calculated critical differences $CD_{0.95}$

| Amine concentration in the condensate, mg/dm³ | <i>CD</i> _{0.95} , % |
|---|-------------------------------|
| 0.5 | 22.2 |
| 1.0 | 7.7 |
| 1.5 | 5.4 |
| 2.0 | 6.0 |
| 5.0 | 6.8 |
| 10.0 | 4.5 |

results obtained in the IREA, Moscow Power Engineering Institute National Research Center, and Travers Research and Production Firm.

Comparison of the results of the determination of amine concentrations in the condensate within the concentration range 0.5–2.0 mg/dm³ was performed for four concentrations: 0.5, 1.0, 1.5, and 2.0 mg/dm³.

Within the concentration range of 2.0–2.0 mg/dm³, comparison was performed for two concentrations: 5.0 and 10.0 mg/dm³.

Mean value of results and sampling variance $S_{i \text{ abs}}^2$, were computed for each laboratory; and the hypothesis of homogeneity of the absolute value of the sampling variance according to Cochran's criterion have also been corroborated.

The calculated mean values, sampling variance, and g_{max} are presented in Table 7. The results obtained at the MPEI and Travers were compared with the results obtained at the IREA. As the acceptable divergence between the results of two laboratories we took the critical difference $CD_{0.95}$ calculated by the formula $CD_{0.95} = 2\sqrt{\Delta}$.

The calculated critical differences $CD_{0.95}$ are presented in Table 8.

The results of the comparison show that the divergence between the results of different laboratories fall inside the $CD_{0.95}$ norm. In other words, the results of all the laboratories are acceptable. This conclusion together with the intermediate data was entered in the $Data\ Processing$ subcategory of the CALS project, for the example of the photometric analysis for amines in an aqueous mixture of dimethylethanolamine, cyclohexylamine, and morpholine.

REFERENCES

- Saaksvuori, A. and Immonen, A., Product Lifecycle Management, Berlin: Springer, 2010, 3rd ed.
- Bakaev, V.V., Sudov, E.V., and Gomozov, V.A., *Informatsionnoe obespechenie, podderzhka i soprovozhdenie zhiznennogo tsikla izdeliya* (Informational Support and Maintenance of Product Lificycle), Moscow: Mashinostroenie-1, 2005.
- 3. Norenkov, I.P. and Kuzmik, P.K., *Informatsionnaya* podderzhka naukoemkihh izdelii (Informatioal Support of High-Tech Products), Moscow: MGTU im. N.E. Baumana.
- 4. Mikhailov, V., Nats. Ekon. Gazeta, 2006, no. 10, p. 928.
- 5. Bessarabov, A.M., Efimova, V.P., and Demyanyuk, A.Yu., *Pribory Avtomatizats.*, 2002, no. 10, pp. 48–54.
- 6. CALS standards // http://www.cals.ru.
- 7. Bessarabov, A.M., Afanasev, A.N., Efimova, V.P., and Ryabenko, E.A., *Khimiya Rynok*, 2001, no. 3, pp. 43–45.
- 8. Bessarabov, A., Kvasyuk, A., Kochetygov, A., and Ivanov, M., *J. Mater. Science Eng., Part A*, 2011, vol. 1, no. 3, pp. 374–379.
- 9. Bessarabov, A.M., Ivanov, M.Ya., and Kvasyuk, A.V., *Ross. Nanotekh.*, 2012, vol. 7, nos. 1–2, pp. 20–23.
- 10. Shalumov, B.Z., Zhebrovskaya, E.B., Shtrambrand, Yu.M., and Bessarabov, A.M., *J. Mend. Chem. Soc.*, 1985, no. 3, pp. 360–361.
- 11. Shalumov, B.Z., Bomshtein, E.V., Bessarabov, A.M., and D'yakova, V.V., *J. Mend. Chem. Soc.*, 1985, no. 6, pp. 574–575.
- 12. Bessarabov, A.M., Vendilo, A.G., Trokhin, V.E., Popov, A.K., Kazakov, A.A., Ryabenko, E.A., and Kochetygov, A.L., *Pribory Sistemy: Upravl., Kontrol, Diagn.*, 2011, no. 2, pp. 1–7.
- 13. Trokhin, V.E., Trynkina, L.V., Kazakov, A.A., and Bessarabov, A.M., *Vestn. Saratov. Gos. Tekh. Univ.*, 2011, no. 4 (62), issue 4, pp. 253–257.
- 14. Bessarabov, A.M., Lobanova, A.V., and Bulatitskii, K.K., *Khim. Prom–st' Segodnya*, 2012, no. 12, pp. 48–55.
- 15. Trokhin, V.E., Vendilo, A.G., Bessarabov, A.M., Kazakov, A.A., and Stepanova, T.I., *Chem. Petrol. Eng.*, 2012, vol. 48, nos. 5–6, pp. 271–277.
- Bessarabov, A.M., Eme'yanova, O.N., Stepanova, T.I., Polyakov, A.V., Zaikov, G.E, and Stoyanov, O.V., *Vestn. Kazan. Tekhnol. Univ.*, 2012, vol. 15, no. 12, pp. 83–88.
- 17. Vendilo, A.G., Bessarabov, A.M., Kovaleva, N.E., Popov, K.I., Stepanova, T.I., and Stoyanov, O.V., *Vestn. Kazan. Tekhnol. Univ.*, 2012, vol. 15, no. 21, pp. 158–163.

- 18. Stepanova, T.I., Kharlamov, A.V., Bondarenko, N.A., and Bessarabov, A.M., *Khim. Prom-st' Segodnya*, 2013, no. 3, pp. 45–50.
- 19. Bessarabov A., Zhekeyev M., Sandu R., Kvasyuk, A., and Stepanova, T., *Chem. Eng. Trans.*, 2012, vol. 26, pp. 513–518.
- 20. Glushko, A.N., Bessarabov, A.M., and Stepanova, T.I., *Naukoemk. Tekhnol.*, 2013, vol. 14, no. 3, pp. 74–80.
- 21. Bessarabov, A.M., Glushko, A.N., Stepanova, T.I., Lobanova, A.V., Zaikov, G.E., and Stoyanov, O.V., *Entsikl. Inzhenera-Khimika*, 2013, no. 7, pp. 51–58.
- 22. GOST (State Standard) 2.114-95: Unified System of

- Design Documentation. Technical Specifications, 2003.
- 23. RMG (Recommendations on Interstate Standardization) 61-2010: State System of the Provision of the Uniformity of Measurements. Accuracy, Correctness, and Precision of the Procedures of Quantitative Chemical Analysis. Assessment Methods, 2012.
- 24. Bulatitskii, K.K. and Lobanova, A.V., *Metody Otsenki Sootvetstv.*, 2010, no. 11, pp. 44–47.
- MU (Methodical Guidelines) 6/113-30-1983: Regulated Measurement Accuracy Parameters in the Measurement Procedure Specified in the Chemical Product Documentation, 1985.